

NJWEA 11th Annual Conference

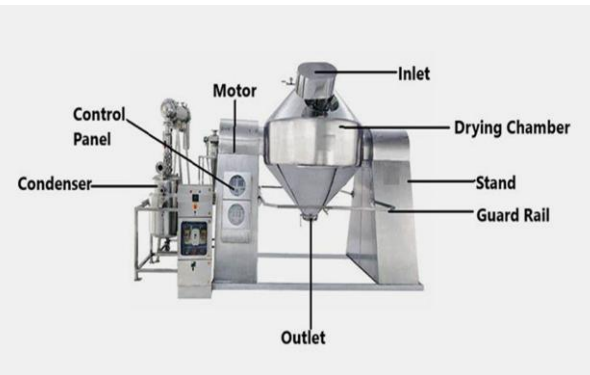
Session: Air Pollution & GHG

Atlantic City, NJ May 12, 2026

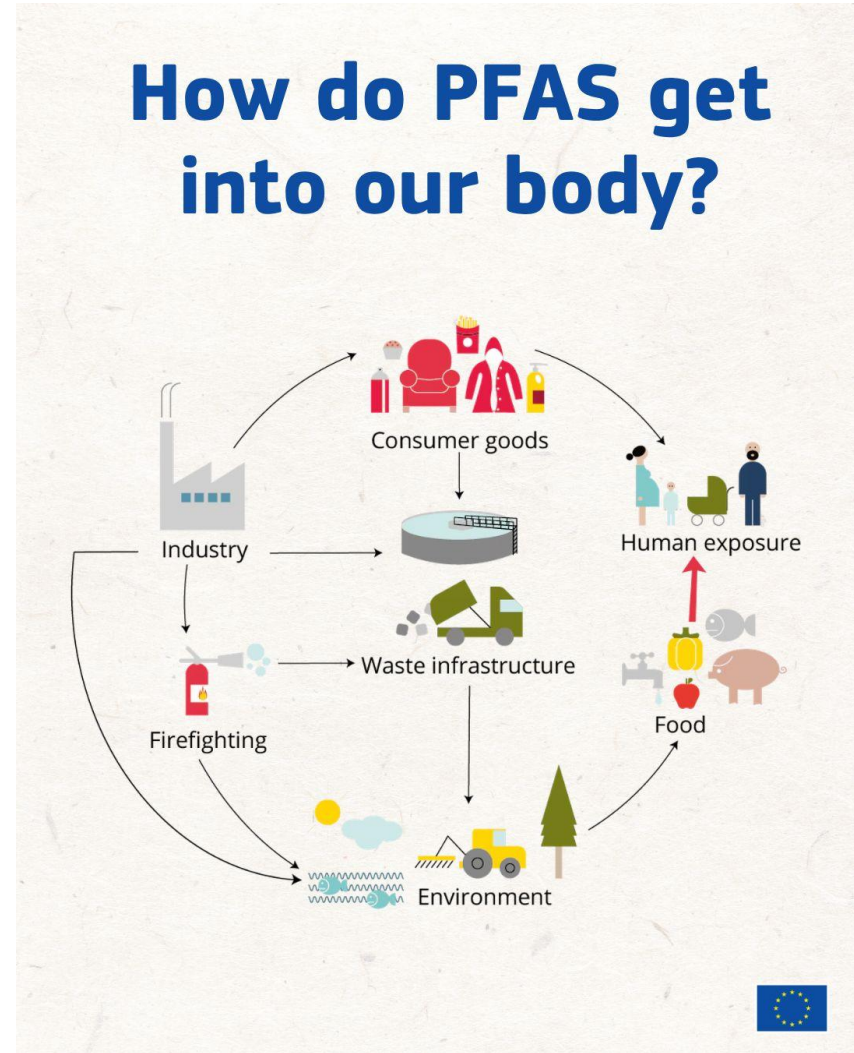
SoPE PROCESS CLEANS UP PFAS FROM BIOSOLIDS: GAS EMISSIONS IMPLICATIONS

By: Edward Weinberg, P.E., President, ESSRE Consulting, Inc.

Desert beetle



PFAS Many Uses Perpetuates the PFAS Cycle

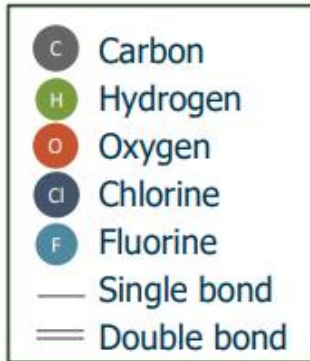


Thousands of PFAS Compounds

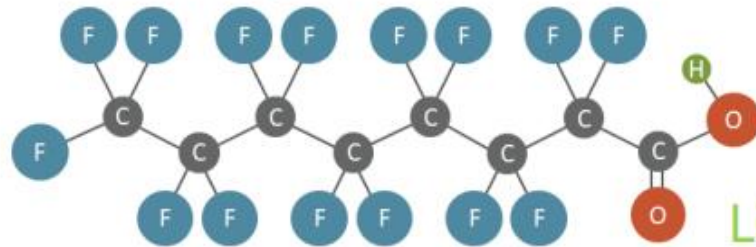
PFAS Chemistry

Perfluoroalkyls: Chain of only C and F atoms

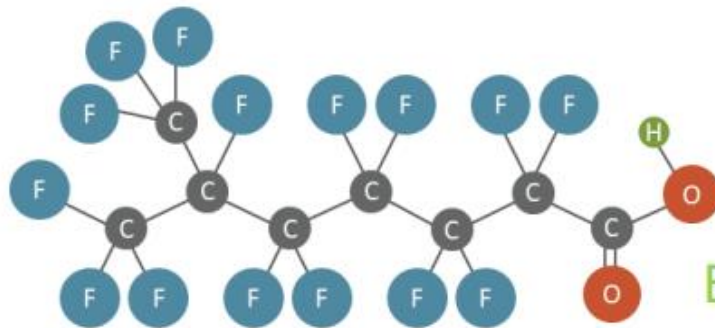
PFCAs (perfluoroalkyl carboxylates)



PFOA (perfluorooctanoic acid)



Linear isomer



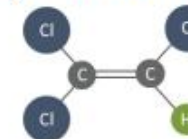
Branched isomer
(1 example)

Methane



Alkyl: only C and H atoms

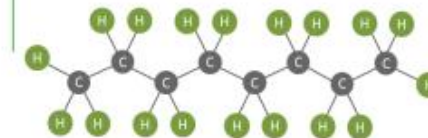
TCE (trichloroethene)



Benzene

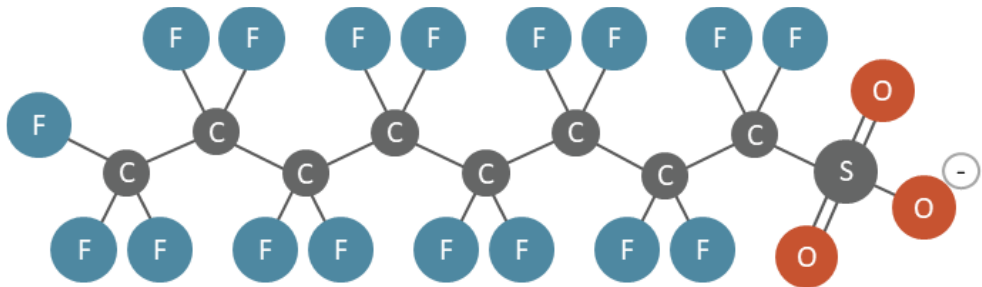


Octane



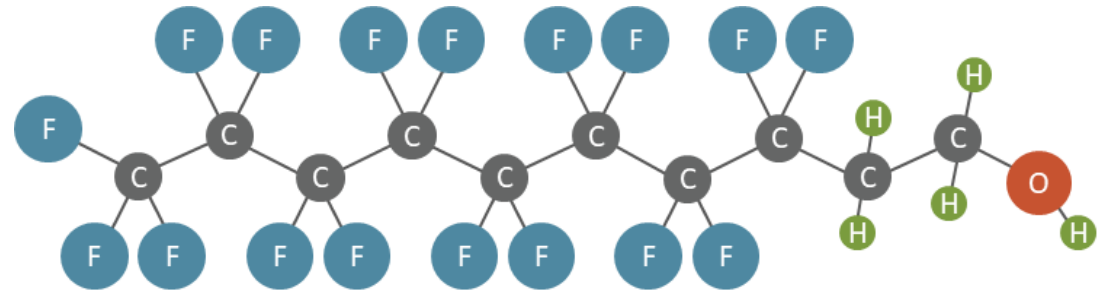
Thousands of PFAS Compounds

PFASs (perfluoroalkane sulfonates)



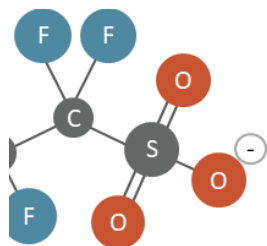
PFOS (perfluorooctane sulfonate)

Polyfluoroalkyl Substances

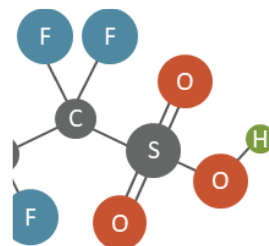


Partially fluorinated, at least one Carbon has a non-fluorine atom. Example, 8:2 Fluorotelomer alcohol (8:2 FTOH)

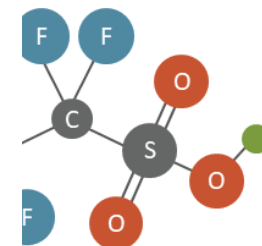
PFAS Compounds Exist in Various States



Anion



Acid



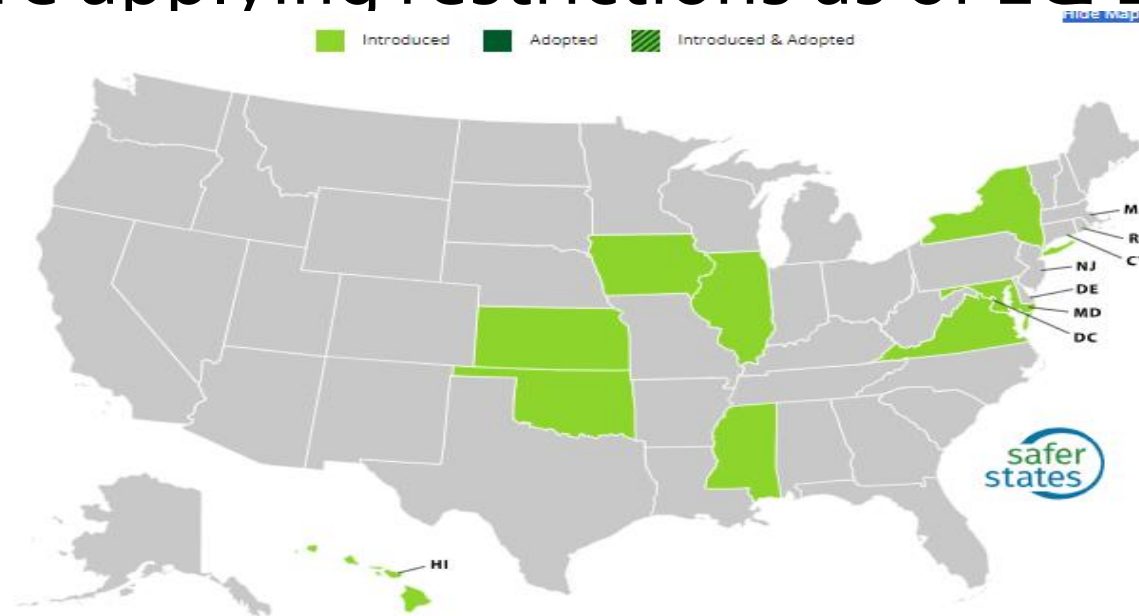
Salt

PFAS “Forever” – Enormous Liabilities

- EPA has finalized maximum contaminant levels for six PFAS compounds
- EPA's Superfund designation of PFOA and PFOS as hazardous substances under CERCLA
- Top 12 PFAS Producers: 3M, AGC, Archroma, Arkema, BASF, Bayer, Chemours, Daikin, Honeywell, Merck, and Solvay
 - Cumulative Annual Profits: **\$4B**
 - 3M's settlement case **\$12.5 B** (drinking water)
- UK Environment Agency estimates remediation costs for between 2,900 and 10,200 high-risk sites in England at between **£31 billion and £121 billion**
- **PFAS Destruction TAM Estimates: Global \$400B; US Biosolids \$36.4 (23,900 WWTP)**

Foreseeing the PFAS Biosolids Future: Banning Land Applications

- ME, CT, MI have fully banned or restricted land application
- MI, NH, MA require biosolids testing
- Several states are applying restrictions as of 1Q 2026:



SoPE Patent –Solid Organic Phase Extraction

- The SoPE process repartitions hydrophobic compounds (PCBs, Hg, PFAS) from solids, soils, sediments, biosolids or spent GAC/IXR onto another more hydrophobic solid phase material
- Why is PFAS hydrophobic?
 - Strong Carbon-Fluorine (C-F) Bond
 - F makes it highly Electronegative
 - Resistant to dissolving in water
 - Caused by the fluorinated “tail”

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United States Patent [19]
Shannon et al.



US005242598A

[11] **Patent Number:** 5,242,598

[45] **Date of Patent:** Sep. 7, 1993

[54] **SOLID PHASE EXTRACTION**
[75] **Inventors:** Michael J. R. Shannon, Plainsboro, N.J.; Burt D. Ensley, Newtown, Pa.
[73] **Assignee:** Envirogen, Inc., Lawrenceville, N.J.
[21] **Appl. No.:** 936,795
[22] **Filed:** Aug. 28, 1992

Related U.S. Application Data

[63] Continuation-in-part of Ser. No. 744,902, Aug. 14, 1991, abandoned.
[51] **Int. Cl.⁵** B01D 11/02; B01D 11/00; B01D 15/00; B08B 7/04
[52] **U.S. Cl.** 210/690; 210/692; 210/747; 210/908; 210/909; 210/924; 134/7; 134/25.1; 134/42; 405/128; 196/14.52; 209/5; 209/173
[58] **Field of Search** 134/6, 7, 25.1, 20, 134/42; 209/5, 9, 172, 172.5, 173; 196/14.52; 405/128, 129; 208/425, 428; 210/747, 691, 692, 693, 908, 909, 690, 924

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Column Chromatographic Determination of Mirex, Photomirex, and Polychlorinated Biphenyls in Lake Sediments, by Alfred S. Y. Chau and Ludwig J. Babjak, Chau & Babjak: J. Assoc. Off. Anal. Chem. (vol. 62, No. 1, 1979) pp. 107-113.
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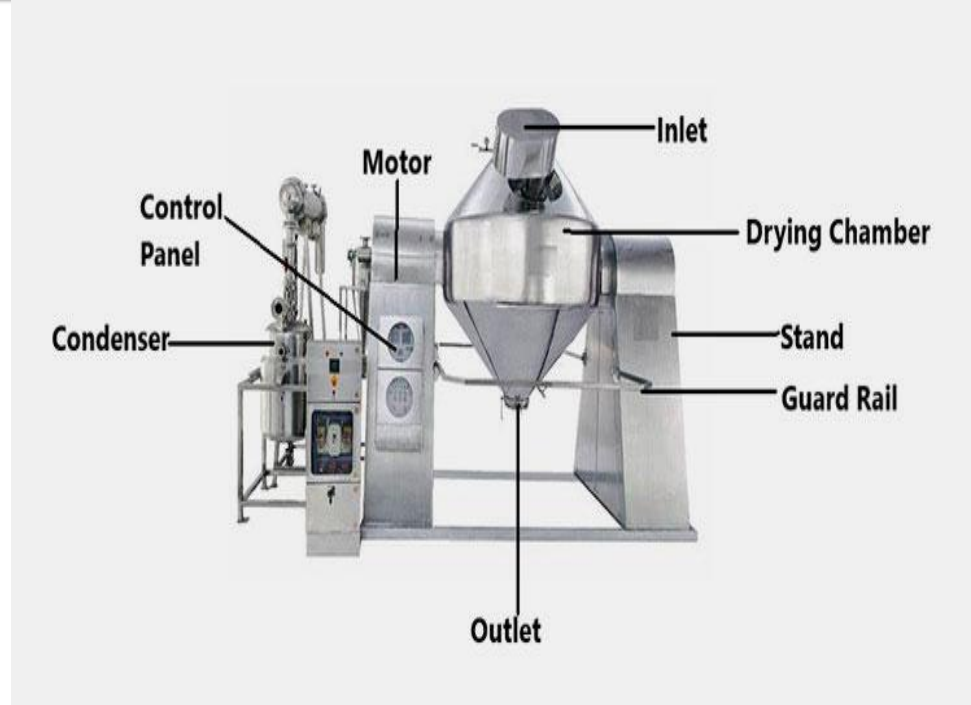
Primary Examiner—Thomas M. Lithgow
Attorney, Agent, or Firm—Mathews, Woodbridge & Collins

[57] **ABSTRACT**

Hydrophobic contaminants are removed from a particulate matter by contacting the particulate matter with a solid organic phase. The hydrophobic contaminants have greater affinity for the solid organic phase than for the particulate matter and repartition to the solid organic phase. The solid organic phase is then separated from the particulate matter. A typical embodiment utilizes polystyrene particles to remove polychlorinated biphenyls from soil.

14 Claims, No Drawings

SoPE Process Steps and Materials



Simple Batch Operation:

- Add ingredients thru INLET
 - 1) Biosolids + Solvent (mix)
 - 2) Water (mix)
 - 3) Extractant Material
- Mix all for designated time and recover solvent
- Discharge thru OUTLET
- Separate Extractant Material from Biosolids (not shown)

Two Solid Organic Phase Extractants: **Expanded Polystyrene Beads & Rubber Crumb/Tire Tread**



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SoPE PCB Soil Results

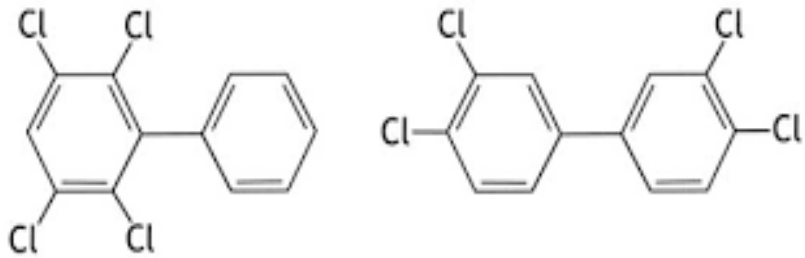
Solvent	PCB REMOVAL (%)	
	Process #1	Process #2
Methanol	54.8	58.1
Acetone	65.8	97.5
Dioxane	89.5	96.2
Ether/methanol	56.2	93.7
Ether/acetone	55.6	80.7

Process #2 follows steps shown in previous slide

Source: US Patent No. 5,242,598

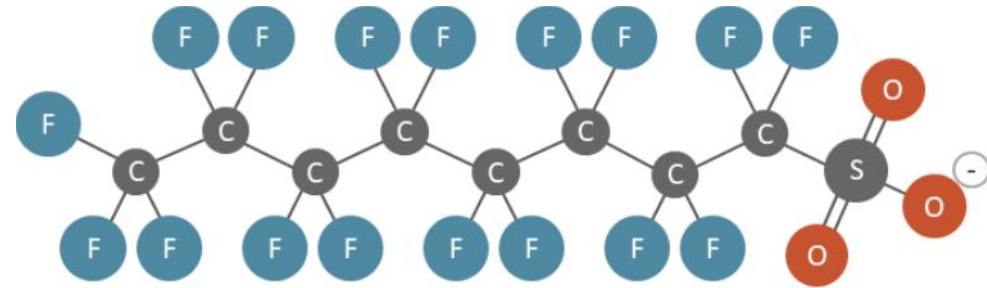
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PFAS vs Polychlorinated Biphenyls (PCBs)




PCBs: Highly chlorinated the more hydrophobic

Only Cl and H atoms = 209 PCB Congeners

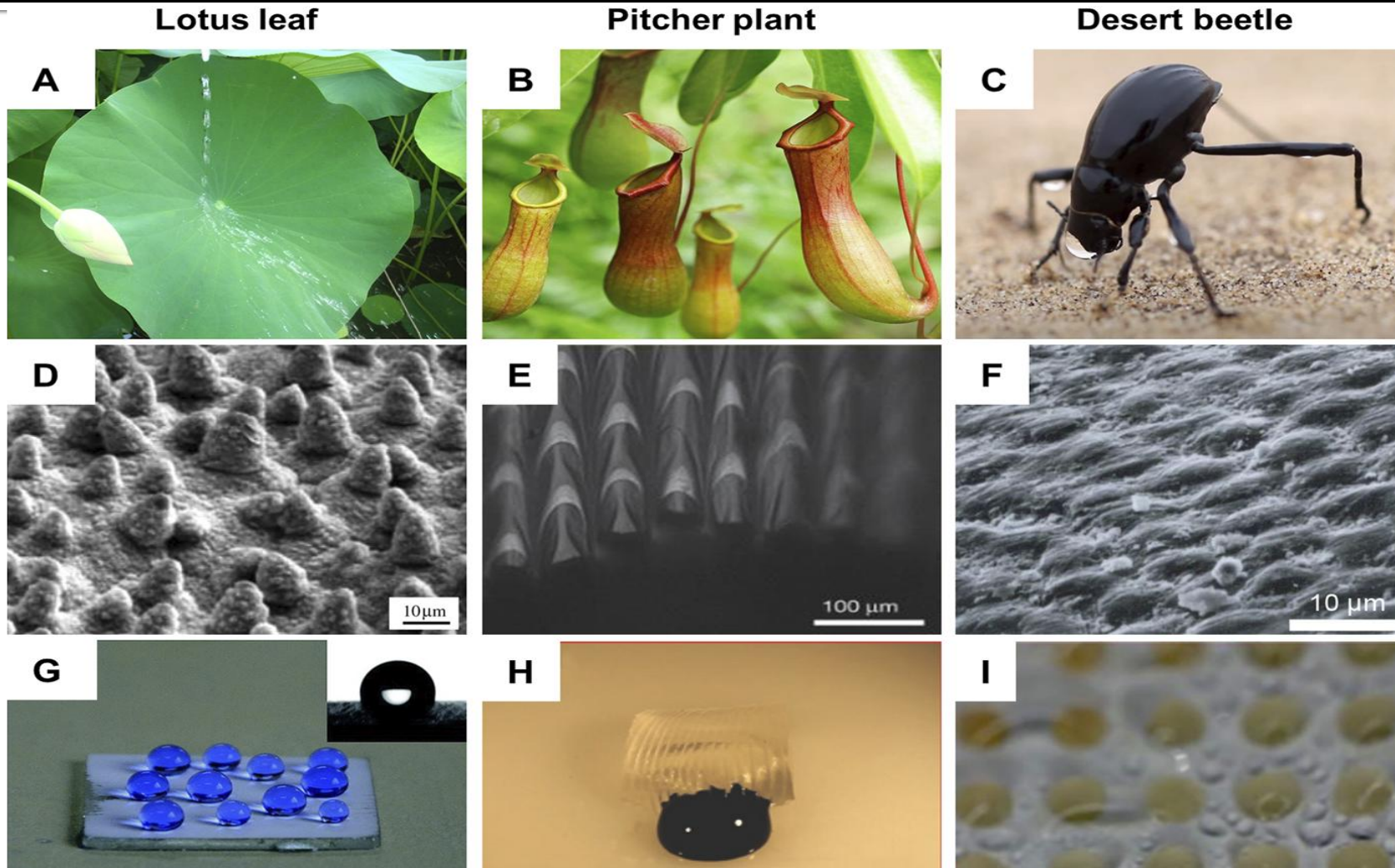


PFAS: Longer C chain the more hydrophobic

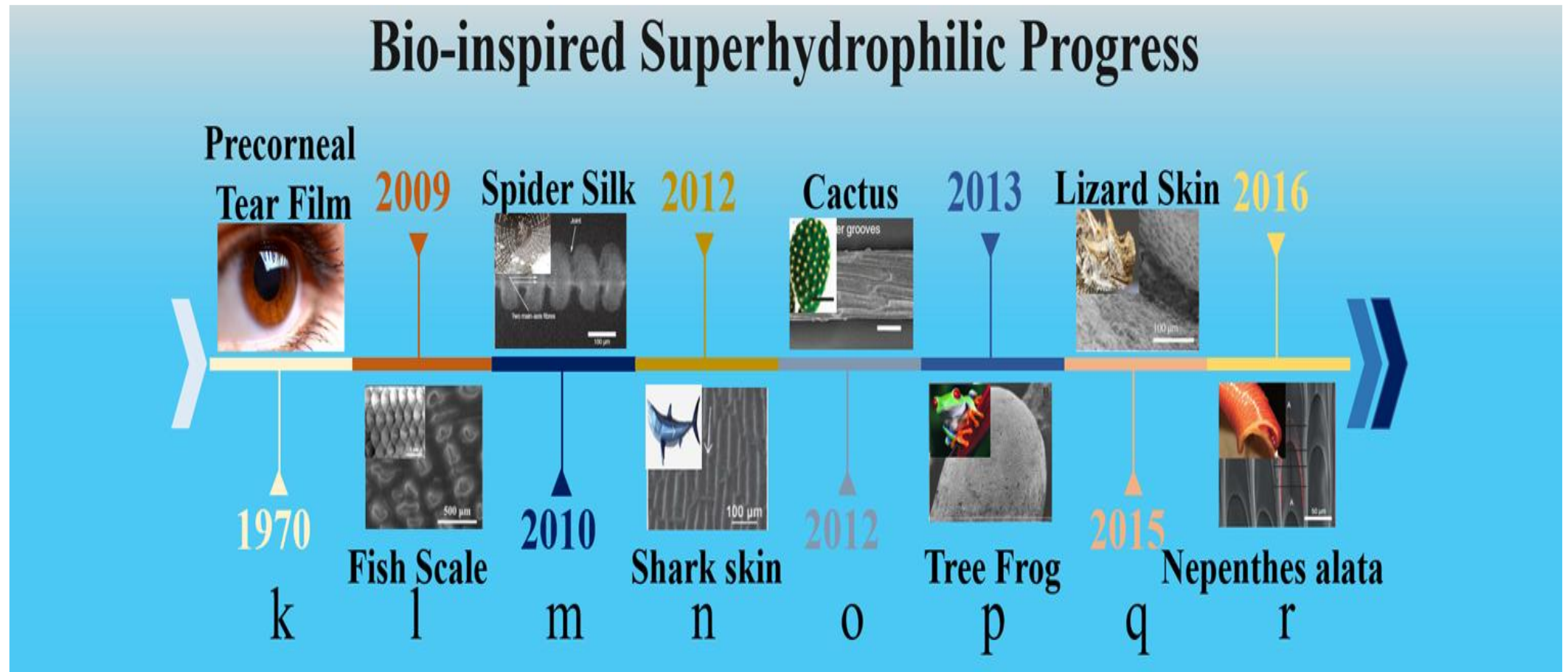
PFAS Hydrophobic Repartitioning Challenges

- PFAS Compared to PCBs
 - Hydrophilic Tail
 - PPT vs. PPM Concentrations (lower driving force & probabilistic contact upon mixing)
 - PFAS lower partitioning coefficients (K_d)
 - Remediation of mix of PFAS compounds
 - Biosolids co-contaminants vs. PCB contaminated soils
- Potential Answer:  SoPE

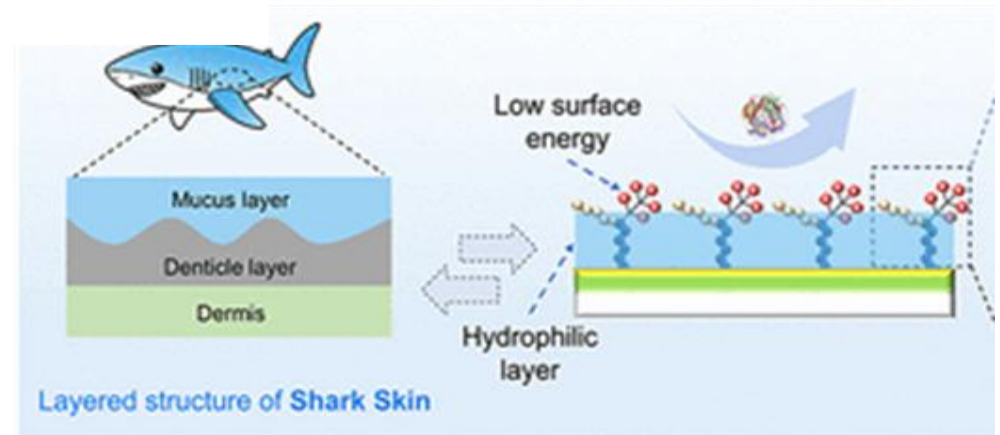
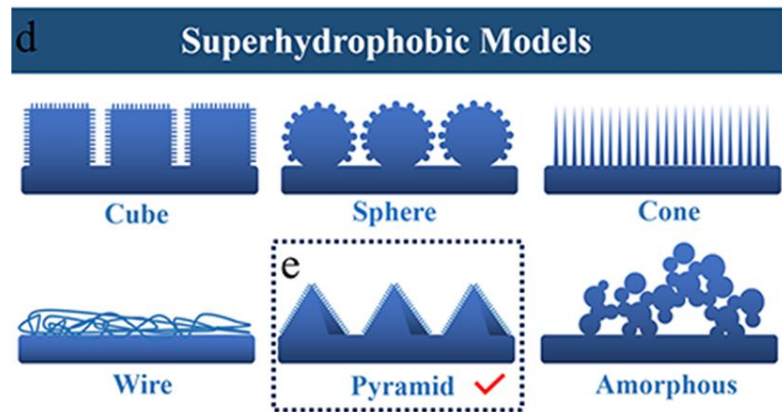
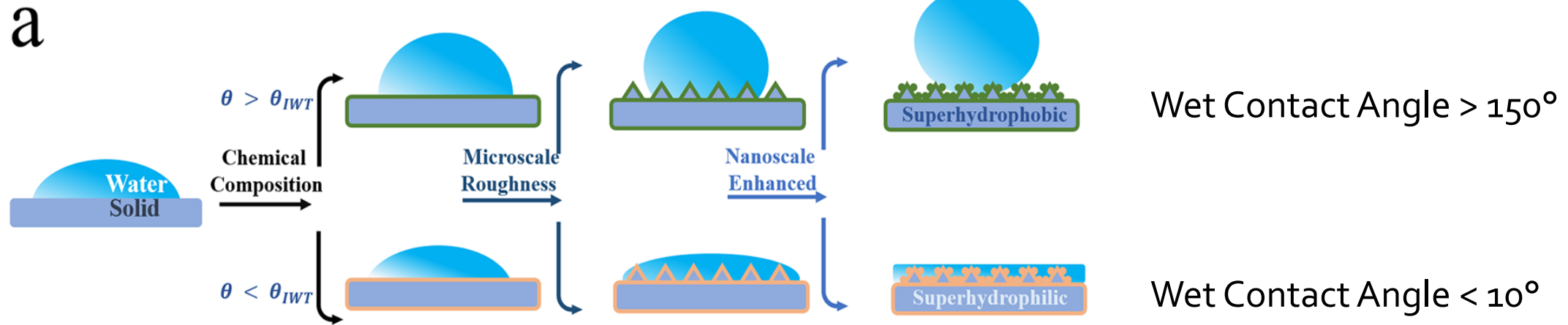
Superhydrophobic Biomimicry



Superhydrophilic Biomimicry



"Super" SoPE - Hydrophobic, Hydrophilic Surfaces



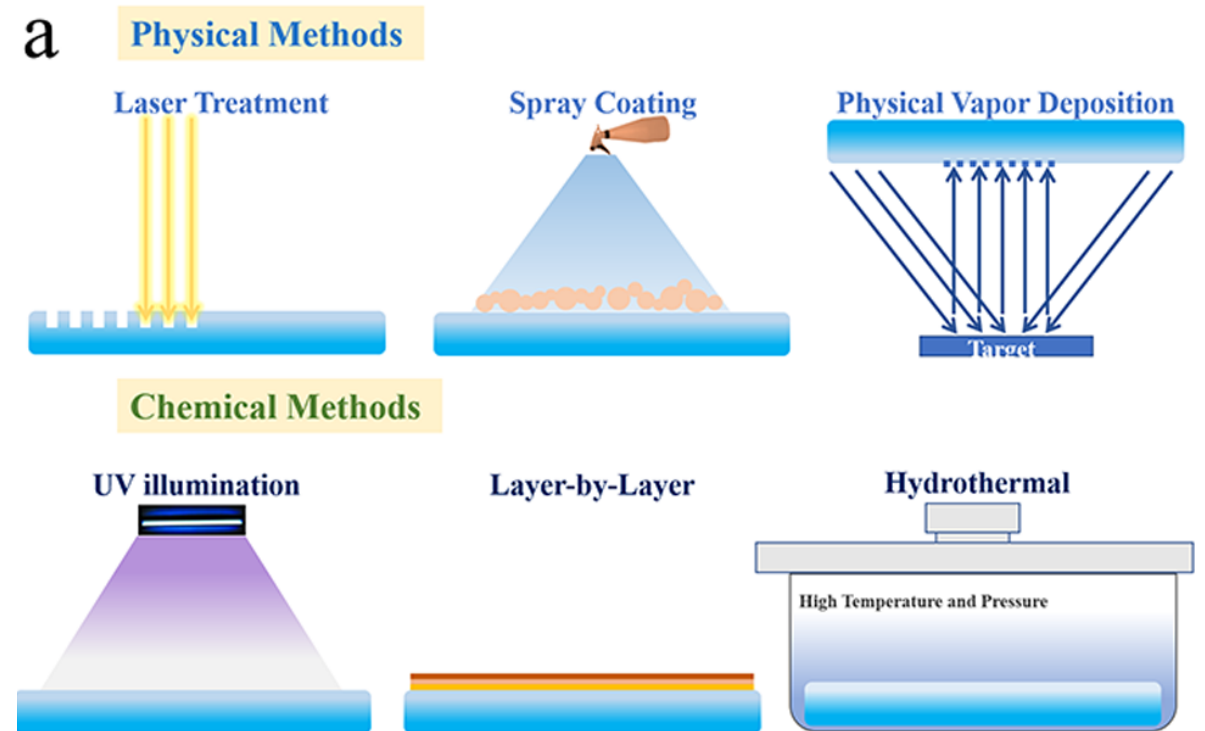
Source: Yifan Si, Zhichao Dong, and Lei Jiang, *Bioinspired Designs of Superhydrophobic and Superhydrophilic Materials*, ACS Cent. Sci. 2018, 4, 1102–1112

Approaches to "Super" Surfaces

- Two Solid Organic Phase Extractants for Consideration: Polystyrene and Rubber

b Hierarchical Micro-Nano Structures

1. Template Replication
2. Photolithography
3. Electrospinning
4. Sol-Gel Processing
5. Other methods



SoPE Complements PFAS Destruction Technologies

- SoPE provides significant material volume reduction
 - EPBs up to 100:1 reduction, 1000 cy biosolids → 10 cy EPB for destruction
- Inversely provides significant PFAS concentration
- EPA guidance document (2024[†]) focuses on the effectiveness of the following emerging destruction technologies from available thermal and non-thermal treatments: mechanochemical degradation, electrochemical oxidation, gasification/**pyrolysis**, and supercritical water oxidation for the destruction of PFAS from solid materials
- Assumed most appropriate destruction technology complementary to SoPE : **PYROLYSIS**
- SoPE flips the pyrolysis script from **biosolids*** to **tire rubber** or **polystyrene**

*Thoma, E. D., Wright, R. S., George, I., Krause, M., Presezzi, D., Villa, V., Preston, W., Deshmukh, P., Kauppi, P., & Zemek, P. G. (2022). Pyrolysis processing of PFAS-impacted biosolids, a pilot study. *Journal of the Air & Waste Management Association*, 72(4), 309-318. <https://doi.org/10.1080/10962247.2021.2009935>

Auto Tire Management via Pyrolysis



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Commercial Waste Tire Pyrolysis Plant

End Products:

Steel separated prior to Pyrolysis

Pyrolysis products: (Liq., Gas, Solid)

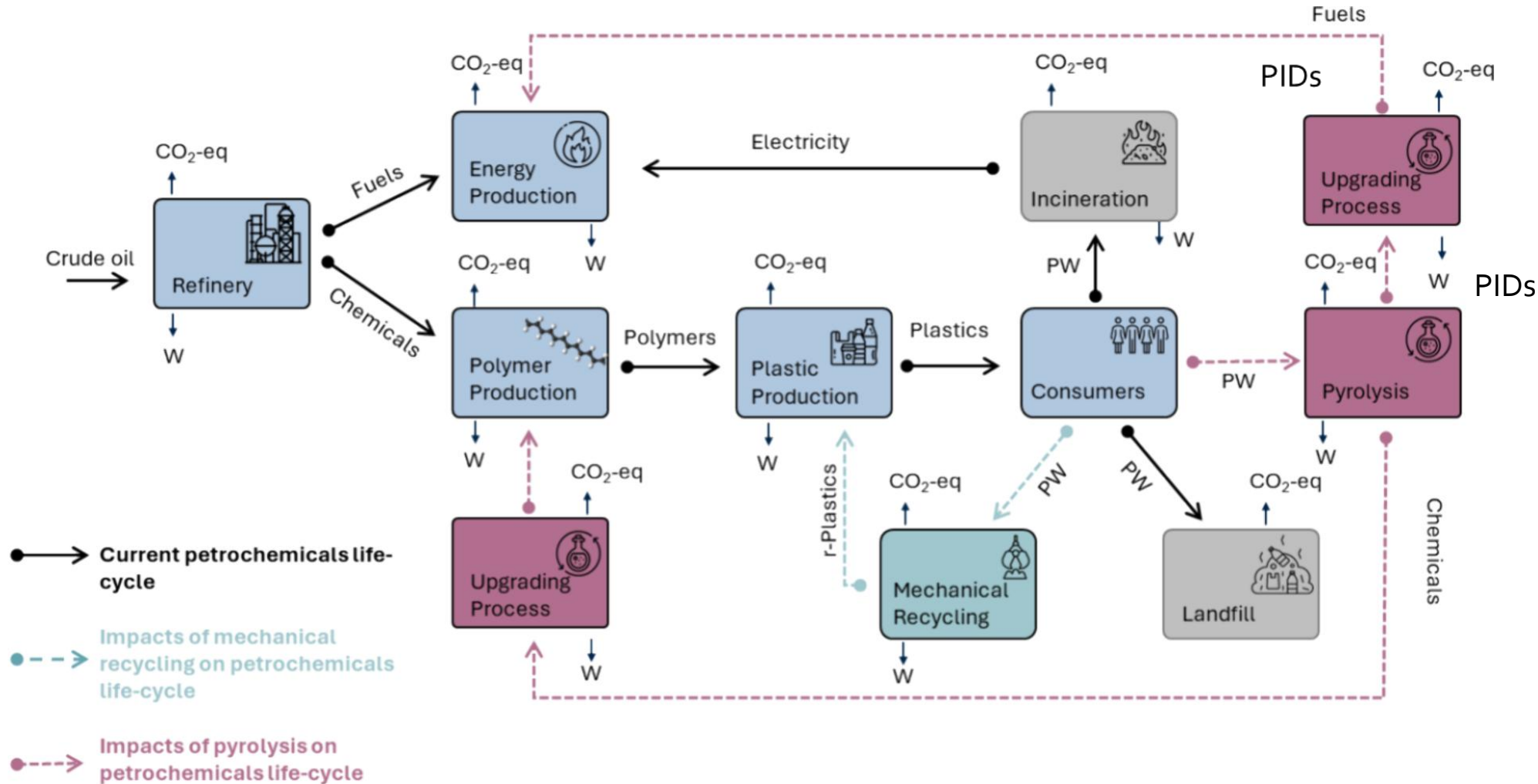
Pyro-oil, Syngas/non-condensable gases, carbon black

PFAS Products of Incomplete

Combustion?? (PICs) Destruction?? (PIDs)

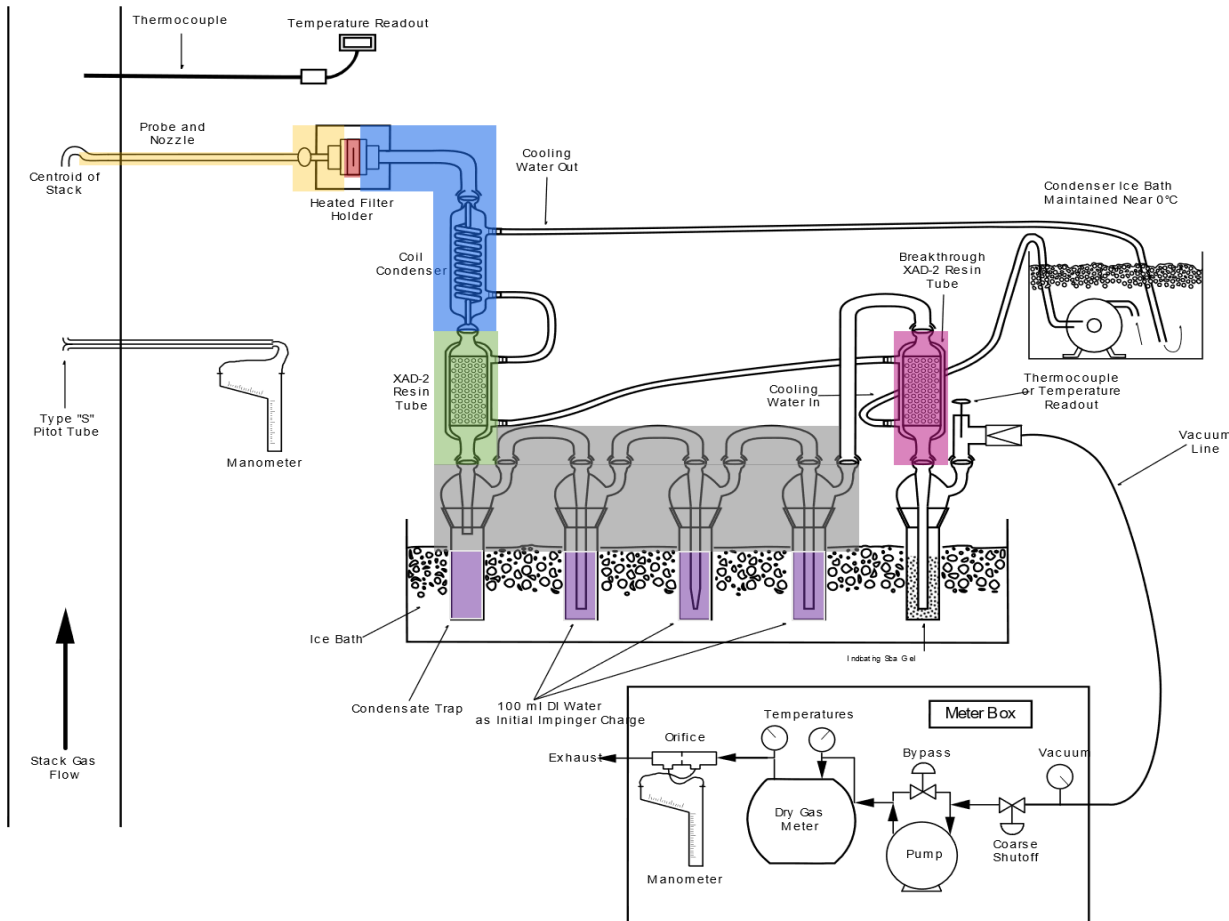
Alternative Source – Playground Tire Crumb???

Plastic Waste Management via Pyrolysis



Pyrolysis of SoPE
 PS provides
 styrene
 monomer **gas** a
 building block to
 make PS and
 other plastics.
Bio-oil and
Biochar

PFAS Gas Emissions Testing



EPA OTM-45: *Measurement of Selected Per- and Polyfluorinated Alkyl Substances from Stationary Sources*, U.S. EPA, January 2021

- List of 49 PFAS compounds are measured.
https://www.epa.gov/sites/default/files/2021-01/documents/otm_45_semivolatile_pfas_1-13-21.pdf

(Newer) **EPA OTM-50:** *Measurement of Selected Per- and Polyfluorinated Alkyl Substances from Stationary Sources*, U.S. EPA, January 2024, Rev 0 2025

- Collection in SS canistirs. Enables analysis of non-polar volatile PFAS thus better characterization of PICs

<https://www.epa.gov/emc/emc-other-test-methods>

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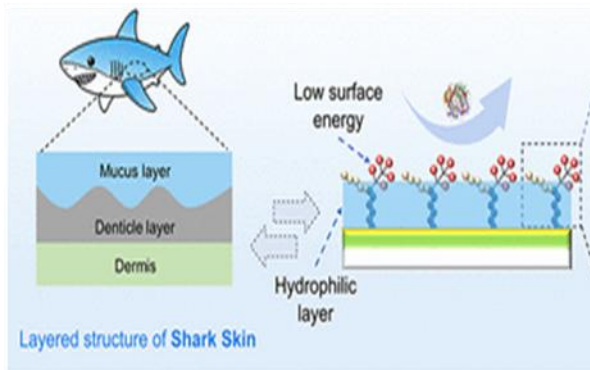
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Questions

